	YEAR 10		YEAR 11
CHEMISTRY 1	Atomic Structure – atoms, history of the atom, structure of the atom, electronic structures and isotopes. The periodic table – Group 1 and 7, electronic structures and explaining trends. Structure and bonding – states of matter, ionic bonding and properties, covalent bonding and properties. Metallic bonding, fullerenes and graphene. Chemical calculations – Relative masses, moles, equations, balanced equations, expressing concentration. Chemical Changes – Reactivity series, Displacement reactions, extraction of metals, making salts, neutralisation and the pH scale. Electrolysis – Changes at electrodes, extraction of aluminum, electrolysis of aqueous solution. Energy Changes – Endothermic and exothermic reactions, reaction profiles and Bond energy calculations 2 x Topic Written Assessments 1 x Unit Exam	PHYSCS 2	Forces in balance— Vectors, scalars, resultant forces, center of mass, parallelogram of forces, resolution of forces. Motion— Speed- distance time graphs, Velocity- time graphs and analysing motion graphs. Force and motion — Forces, acceleration, terminal velocity, momentum and elasticity. Wave properties — Nature of waves, properties of waves, reflection and refraction. Electromagnetic waves— The electromagnetic spectrum and communication. Electromagnetism — Magnetic fields, electric currents, the motor effect.
PHYSICS 1	Energy – Conservation of energy, work done, GPE and KE, energy, power and efficiency. Energy transfer – Conduction, specific heat capacity, heating and insulating buildings. Energy resources – Renewable energy, environment and issues. Electric circuits – Current, charge, voltage, series and parallel circuits. Electricity in the home – Alternating current, cables, wires, plugs and appliances. Molecules and matter – Density, States of matter, internal energy, specific latent heat, gas pressure and temperature. Radioactivity – Atoms, radiation, discovery of the nucleus, alpha, beta, gamma and half life. 2 x Topic Written Assessments 1 x Unit Exam	CHEMISTRY 2	Rates and equilibrium – Collision theory, effect of temperature, concentration, pressure, and catalysts. Equilibrium, and reversible reactions. Crude oil – Hydrocarbons, burning and cracking. Chemical Analysis – Pure substances, mixtures, chromatography and testing gases. The Earth's atmosphere – History of the atmosphere, greenhouse gases, climate change and pollutants. The Earth's resources – Finite and renewable resources, water treatment, extraction of metals, life cycle assessment and recycling.
BIOLOGY 1	Cell structure – Microscopes, cells and specialized cells. Diffusion, osmosis, and active transport. Cell Division – Mitosis and stem cells. Organisation – Tissues, digestive system, enzymes, factors affecting enzymes, Organisation of animals and plants – Blood, vessels, heart, breathing and transport systems in plants. Communicable diseases – Health, pathogens, bacteria, virus, fungi and human defence response. Preventing diseases – Vaccination, antibiotics and painkillers, discovering and developing drugs. Photosynthesis – Rate of photosynthesis, use of glucose, limiting factors. Respirations – Aerobic, anaerobic respiration and metabolism.	BIOLOGY 2	The nervous system – Homeostasis and reflex actions Hormonal coordination – Blood glucose, diabetes, negative feedback, menstrual cycle and fertilitiy. Reproduction— Cell division, meiosis, DNA, inheritance, genetics and disorders. Variation & Evolution—Variation, natural selection, selective breeding, genetic engineering and ethics. Genetics and Evolution—Extinction and fossils, antibiotic resistance bacteria, classification systems. Interdependence—Competition between animal, plants, adaptation, and survival. Ecosystem — Feeding relationships and the carbon cycle. Biodiversity — Human population, pollution, deforestation, global warming and maintaining biodiversity.
	2 x Topic Written Assessment 1 x Unit Exam		2 x Topic Written Assessments 1 x Unit Exam